

Density Functional Theory: Introduction and Applications

André Schleife

Department of Materials Science and Engineering

University of Illinois Urbana-Champaign

October 5th, 2022



The Grainger College of Engineering

IBM-Illinois Discovery Accelerator Institute



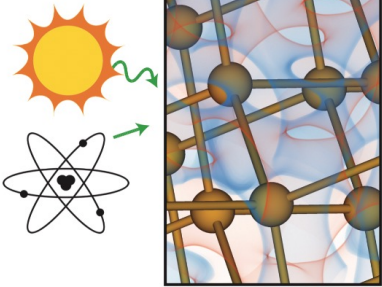
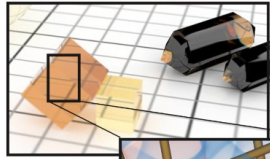
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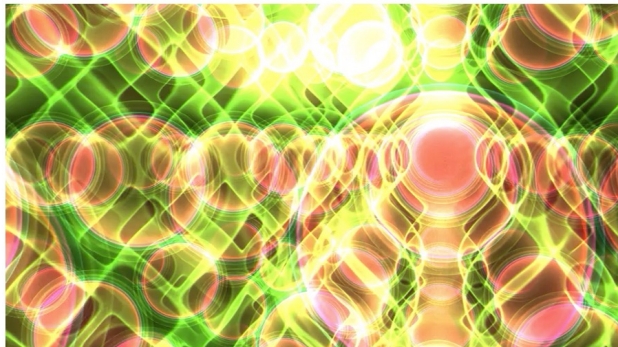
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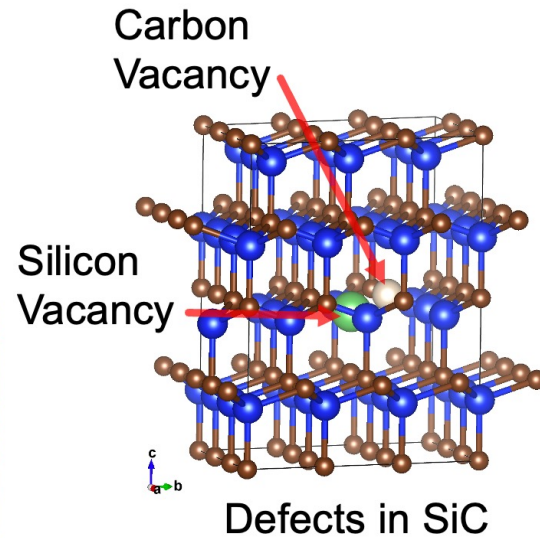
October 5th, 2022



Electronic Excitations



Proton-irradiated MgO

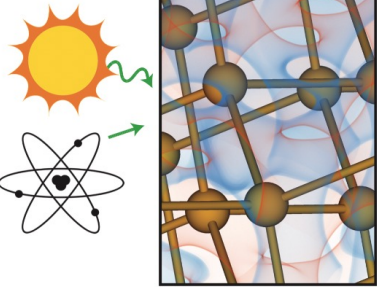
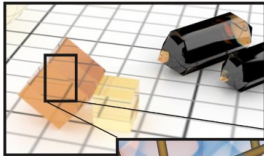


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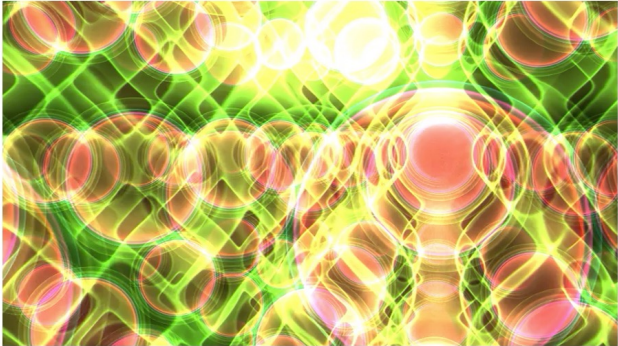
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Schleife
Group
Department of Materials
Science and Engineering

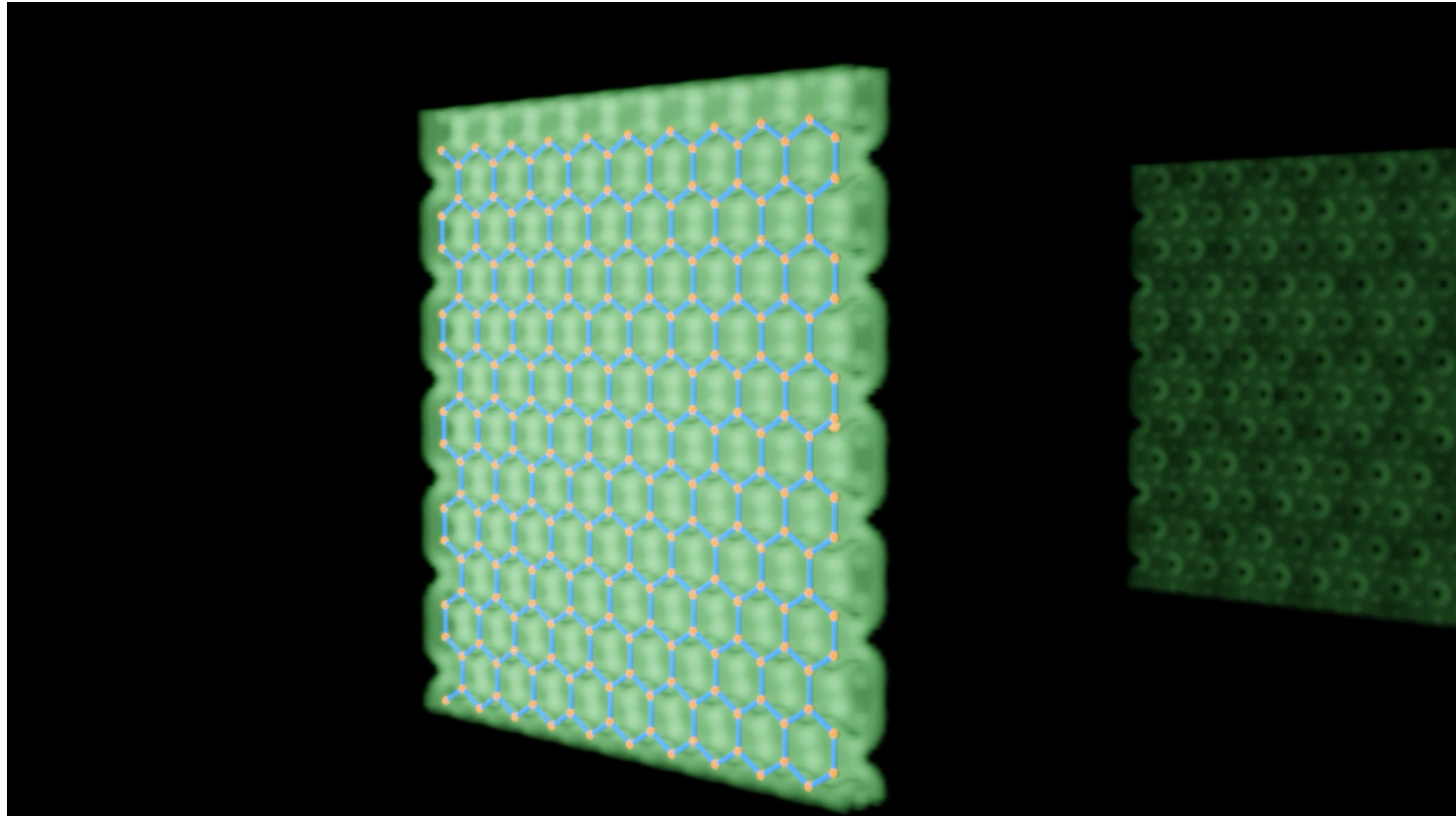


Question: What do you need Density Functional Theory for?

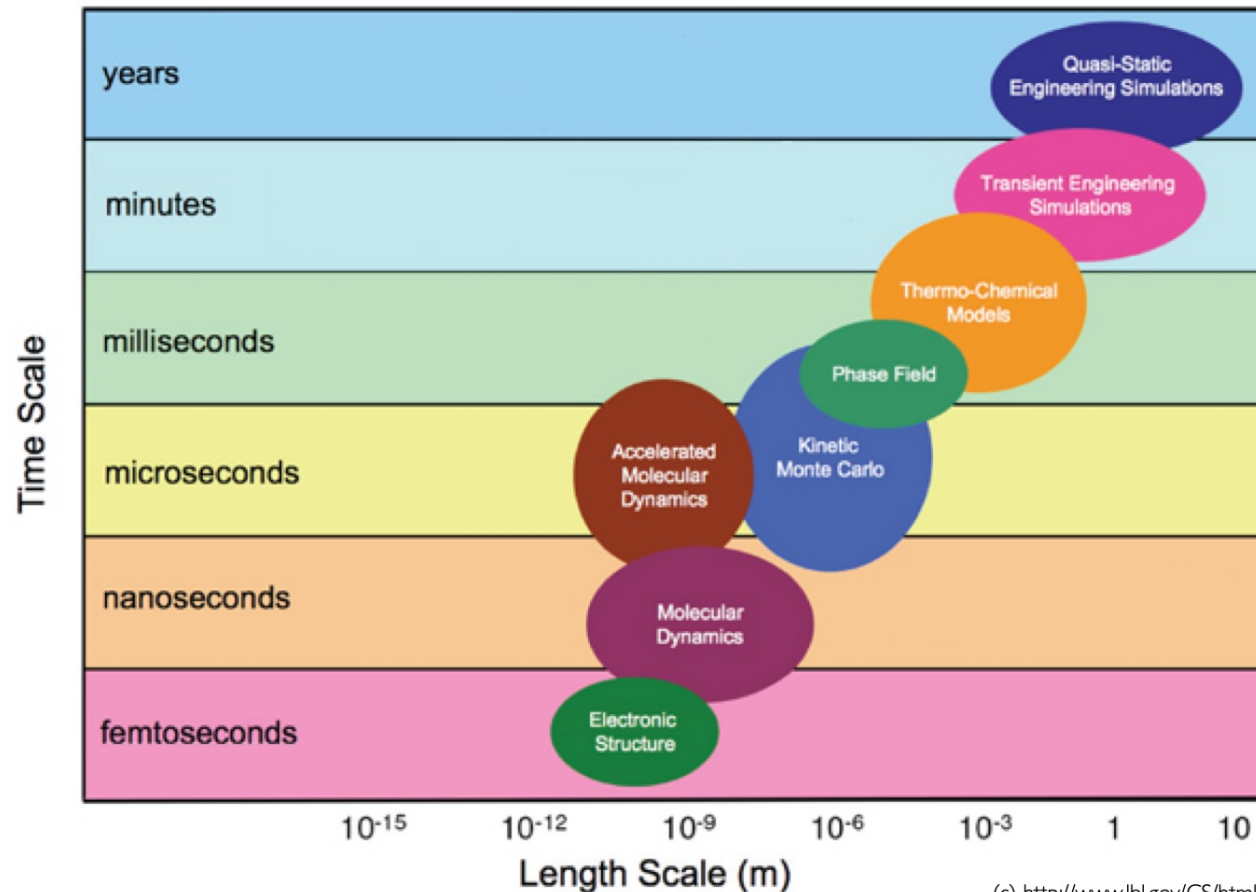
1. Use DFT
2. Develop DFT
3. Mostly to understand other people's work
4. My teaching
5. I am mostly just curious

Overview

- Motivation and Background
- Formulation and Implementation of Density Functional Theory
- Example Applications: Total Energy and Bulk Modulus, Electronic Band Structures



Computational Material Science



(c) <http://www.lbl.gov/CS/html/exascale4energy/nuclear.html>

- Computational cost depends on level of accuracy
- Many different approaches/codes available (depends on goal)
- Density Functional Theory: Microscopic scale

Microscopic Scale: Quantum Mechanics

Today's standard model of materials:

- Materials consist of atoms
- Atoms consist of
 - massive, point-like nuclei (protons + neutrons)
 - surrounded by tightly bound core electrons
 - held together in molecules, liquid, and solids by the bonds formed by valence electrons

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Goal: Solve laws of quantum mechanics for actual materials

- Bonding and structure (structural properties)
- Electronic, magnetic, and optical properties of materials
- Chemistry and reactions

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The general theory of quantum mechanics is now almost complete. The underlying physical laws necessary for the mathematical theory of a large part of physics and the whole of chemistry are thus completely known, **and the difficulty is only that the exact application of these laws leads to equations much too complicated to be soluble.**

P. A. M. Dirac, Proc. R. Soc. Lond. A 1929, pg. 123

Microscopic Scale: Quantum Mechanics

Quantum Mechanics:

1. Write Down Hamilton Operator

$$\hat{H} = \hat{T} + \hat{V}$$

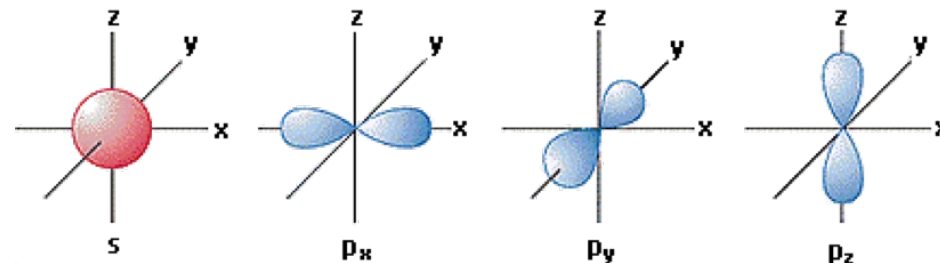
2. Solve Schrödinger Equation to get E and ψ

$$\hat{H} \psi(\mathbf{r}) = (\hat{T} + \hat{V}) \psi(\mathbf{r}) = E \psi(\mathbf{r})$$

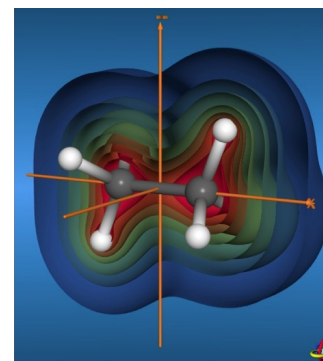
3. Use this solution to compute observables

Probability of finding electron at position \mathbf{r} in space

$$n(\mathbf{r}) = \sum_{\kappa=1}^{\infty} n_{\kappa} |\psi_{\kappa}(\mathbf{r})|^2$$



(c) <https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtjml/Images/hybrid1.gif>



(c) <http://csi.chemie.tu-darmstadt.de/ak/immell/tutorials/orbitals/molecular.html>

Microscopic Scale: Quantum Mechanics

Why is Quantum Mechanics/Solving Schrödinger Equation hard?

Example: Fe atom

- Fe has 26 electrons \Rightarrow wave function has $3 \times 26 = 78$ variables
- store wave function on a grid
- coarse grid of only 10 points along each direction
- to store wave function would require storage of 10^{78} numbers
- single precision 1 number = 4 Bytes
- compare to all data stored worldwide in 2020:

64 zettabyte = 64×10^{21} Bytes

$$\Psi(r_1, \dots, r_{26})$$



(c) https://en.wikipedia.org/wiki/Iron#/media/File:Iron_electrolytic_and_1cm3_cube.jpg

Microscopic Scale: Quantum Mechanics

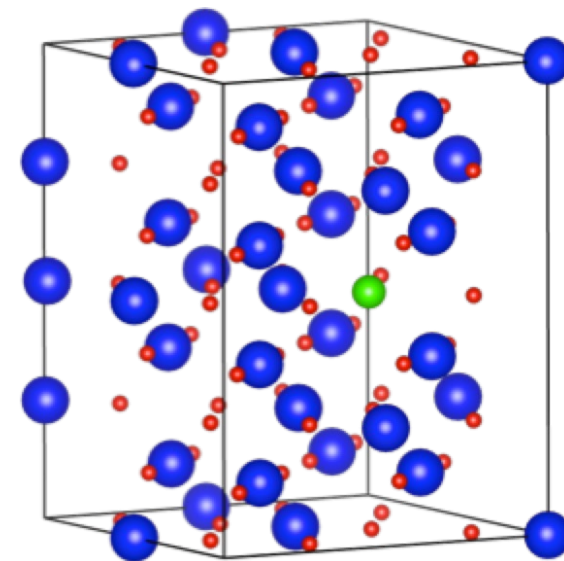
Many-electron Hamiltonian of a material

- Fix ions in their positions: Born-Oppenheimer Approximation
- Leads to Electronic Hamiltonian:

$$\hat{H} = \underbrace{-\sum_i \frac{\hbar^2}{2m_e} \nabla_{\mathbf{r}_i}^2}_{KE} + \underbrace{\sum_i \sum_{j>i} \frac{e^2}{4\pi\epsilon_0 |\mathbf{r}_i - \mathbf{r}_j|}}_{e^- - e^-} - \underbrace{\sum_i \sum_j \frac{Z_i e^2}{4\pi\epsilon_0 |\mathbf{R}_i - \mathbf{r}_j|}}_{e^- - \text{ion}}$$

Solution requires approximations

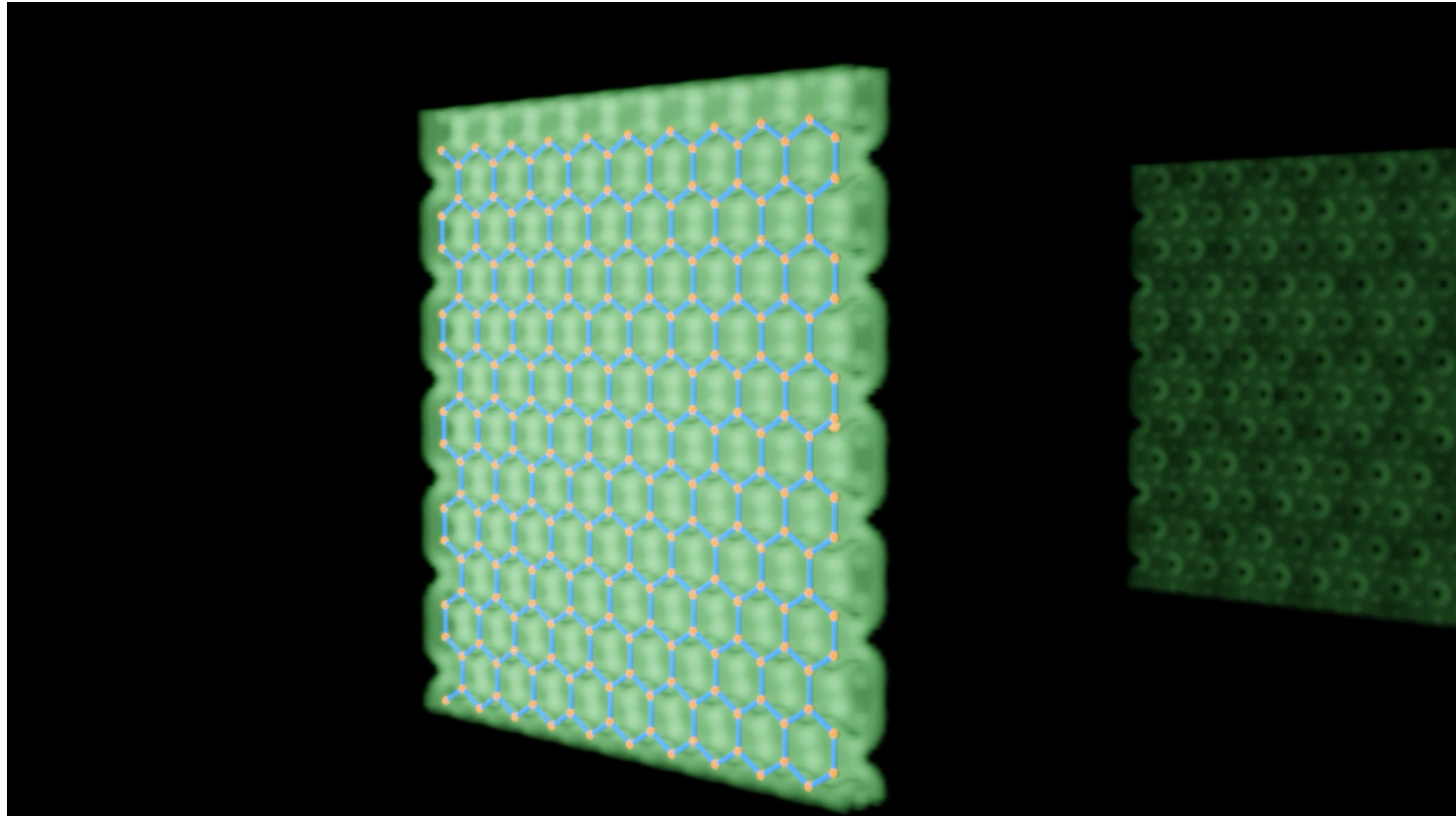
- Historic approaches include Hartree Approximation, Hartree-Fock Approximation, ...
- In general: Balance/tradeoff between accuracy and speed
- Density Functional Theory favorable on that spectrum and is very widespread nowadays



TiO_{2(1-x)}S_{2x}

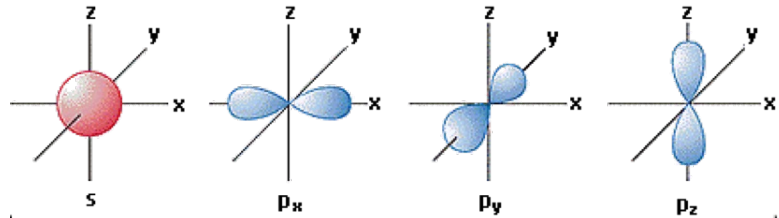
Overview

- Motivation and Background
- **Formulation and Implementation of Density Functional Theory**
- Example Applications: Total Energy and Bulk Modulus, Electronic Band Structures



Density Functional Theory: Formulation and Implementation

- Walter Kohn proved: Ground state energy of the electronic system can be written as functional of electron density (instead of wave function)



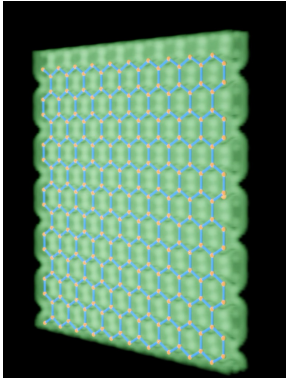
(c) <https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/Images/hybrid1.gif>

$$\Psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_n)$$

Function of $3N$ variables

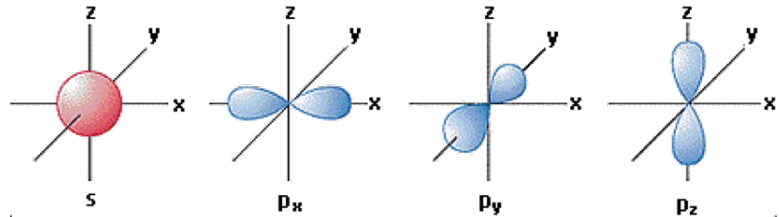
$$n(\mathbf{r})$$

Function of 3 variables



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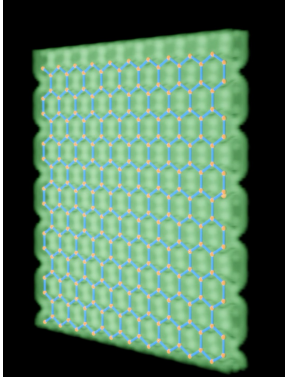
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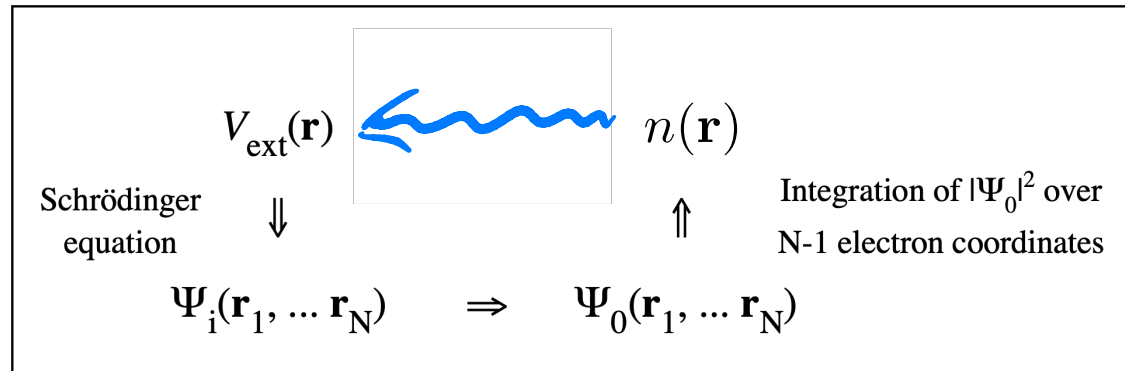
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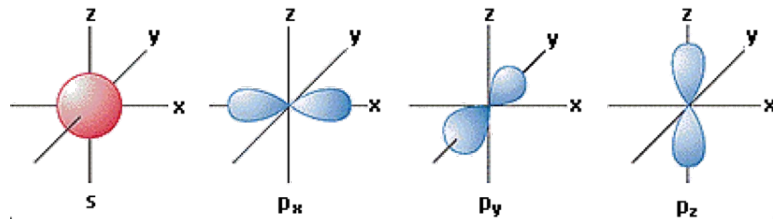
First Hohenberg-Kohn theorem

external potential is a unique functional of density



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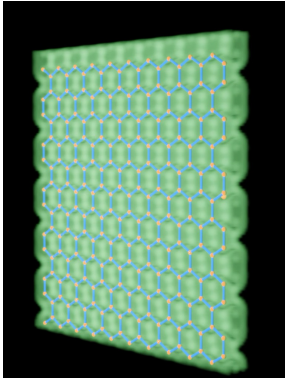
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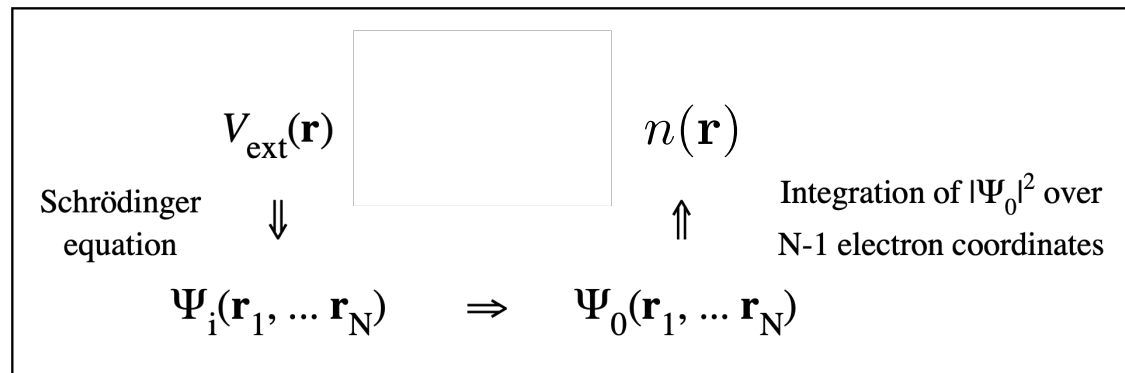
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Function of 3 variables



First Hohenberg-Kohn theorem

external potential is a unique functional of density



Second Hohenberg-Kohn theorem

density that minimizes the total energy is the exact groundstate density

$$E[n(\mathbf{r})] = E_{\text{kin}}[n(\mathbf{r})] + V_{\text{ext}}[n(\mathbf{r})] + V_{ee}[n(\mathbf{r})] \geq E_0$$

Question: Have we made an approximation yet?

1. Yes
2. No
3. I am not sure

Density Functional Theory: Formulation and Implementation

Find the solution in practice: Map to a non-interacting (Kohn-Sham) system

- Reason: We know the kinetic energy of non-interacting electrons

$$n(\mathbf{r}) = \sum_{k=1}^N n_k |\varphi_k(\mathbf{r})|^2 \quad E_{\text{kin}}[n(\mathbf{r})] = -\frac{1}{2} \sum_{i=1}^N \int \varphi_i^*(\mathbf{r}) \nabla^2 \varphi_i(\mathbf{r})$$

- Determine the Hamiltonian of that system, such that the ground-state electron density of interacting and non-interacting system are the same
- But: We replaced the interacting by the non-interacting kinetic energy
- And: We also replaced full electron-electron interaction by the Hartree potential
- Make up for the resulting error by introducing an exchange-correlation functional

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$$\underbrace{\left[-\frac{1}{2} \nabla^2 + V_{\text{H}}(\mathbf{r}) + V_{\text{xc}}(\mathbf{r}) + V_{\text{ext}}(\mathbf{r}) \right]}_{\mathcal{H}_{\text{KS}}} \varphi_i(\mathbf{r}) = \epsilon_i \varphi_i(\mathbf{r})$$

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- Kinetic energy of non-interacting electrons
- Hartree energy (classical electron-electron interaction)
- External potential (Coulomb potential of ions)

$$V_{xc} = \hat{T} - \hat{T}_S + \hat{U} - V_H$$

Density Functional Theory: Formulation and Implementation

Find the solution in practice: Map to a non-interacting (Kohn-Sham) system

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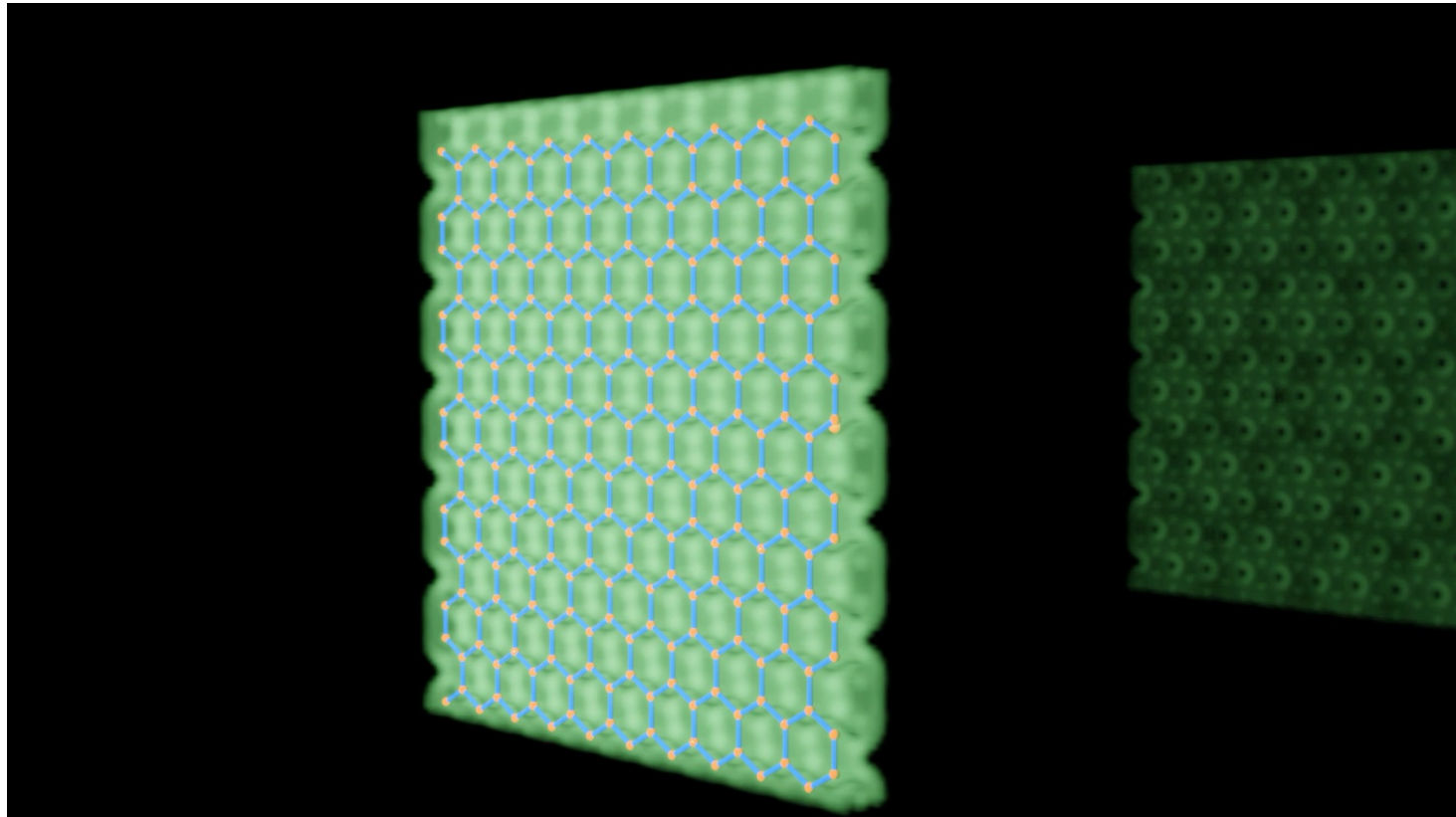
- Kinetic energy of non-interacting electrons
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All quantum effects in exchange-correlation potential

- Exact form not known, but reliable approximations available
- Examples: Local-density approximation (take from free-electron gas of same density), generalized-gradient approximation, self-interaction correction, Hubbard U , hybrid functionals, exact exchange, Van der Waals, Machine-Learned Functionals, ...
- Accuracy: depends on approximation used and material to be studied

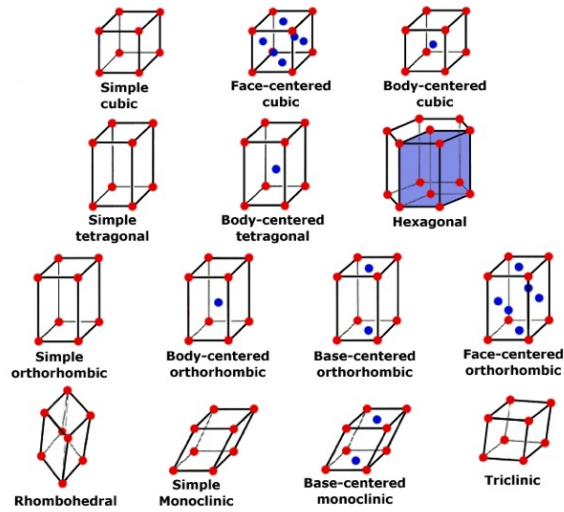
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- Motivation and Background
- Formulation and Implementation of Density Functional Theory
- **Example Applications: Total Energy and Bulk Modulus, Electronic Band Structures**



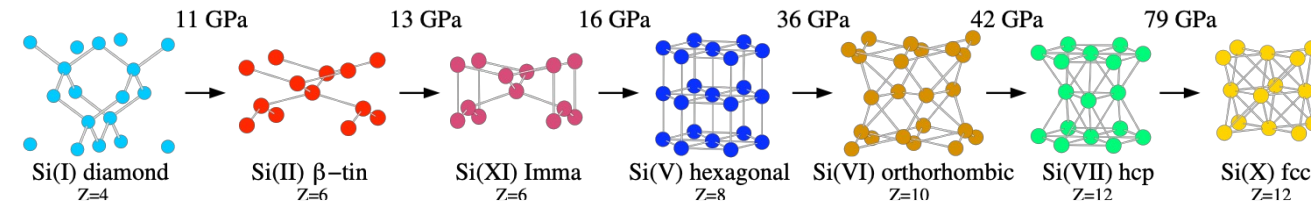
Density Functional Theory: Applications

Predict atomic geometries and crystal structure, e.g. Si

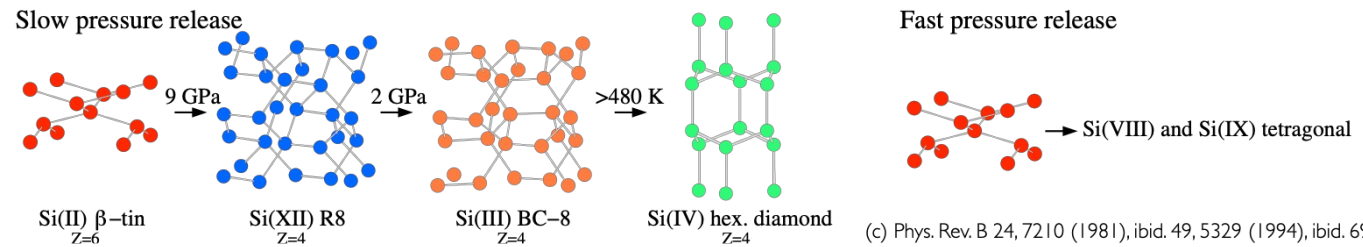


(c) <http://www.seas.upenn.edu/~chem101/sschem/bravais.gif>

Compression



Decompression



(c) Phys. Rev. B 24, 7210 (1981), ibid. 49, 5329 (1994), ibid. 69, 134112 (2004)

Other examples include:

- Ab-initio molecular dynamics
- Vibrational properties
- Defect geometries and energetics
- ...

Density Functional Theory: Applications

Quantum Espresso

- Integrated suite of Open-Source computer codes for electronic-structure calculations and materials modeling at the nanoscale
- Core set of codes, plugins for more advanced tasks and third-party packages
- Contributed to by developers across the world
- Can be run on nanoHUB, including for teaching
- Uses plane wave expansion for Kohn-Sham states and pseudo-potentials for electron-ion interaction

Example I: Total-energy calculations and convergence

- Compute total energies for bulk fcc silicon, change either plane-wave cutoff and/or k-point grid



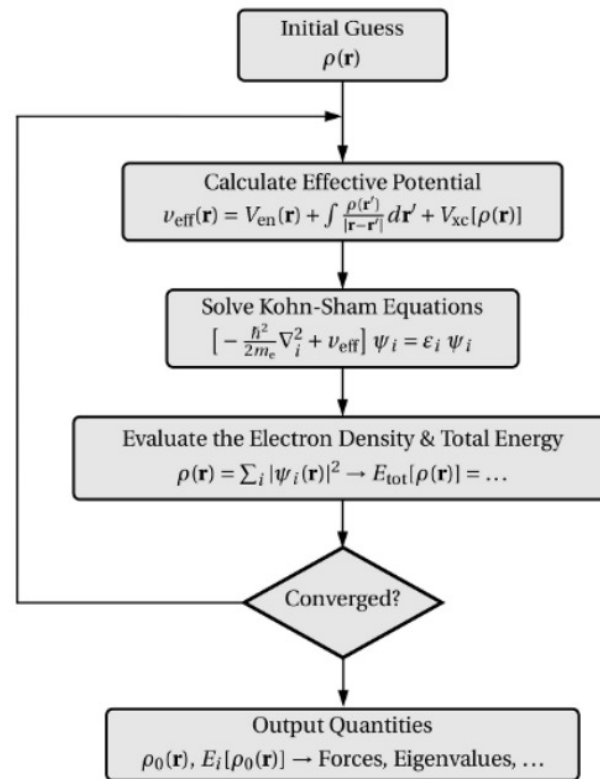
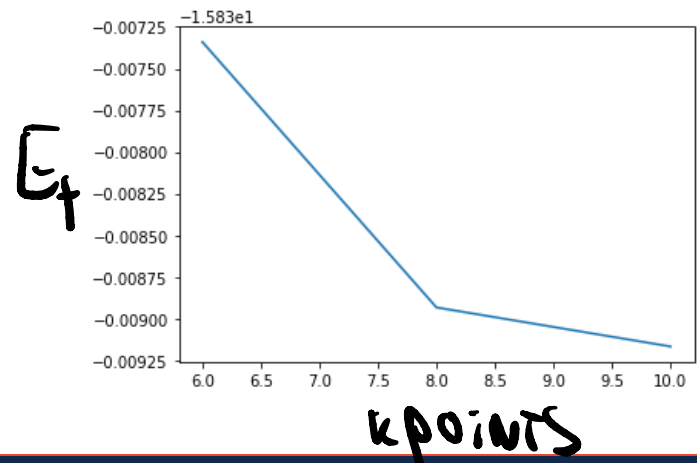
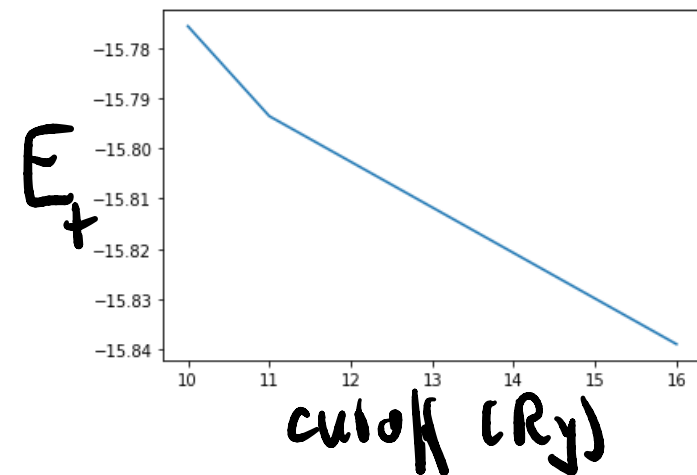
<https://nanohub.org/tools/dftqe>

Density Functional Theory: Applications

Quantum Espresso

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Example I: Total-energy calculations and convergence



<https://nanohub.org/tools/dftqe>

(c) Wolfgang Goes, TU Vienna

Density Functional Theory: Applications

Example II: Bulk modulus

- At constant temperature, bulk modulus relates change in volume with a change in pressure:

$$B = -V \frac{dP}{dV}$$

- Decreasing the volume corresponds to positive external pressure on the material
- Pressure and energy are related via:

$$P = -\frac{dE}{dV}$$

- This leads to the following expression for the bulk modulus:

$$B = V \frac{d^2 E}{dV^2}$$

- Produce an $E(V)$ curve with nanoHUB by varying volume around $0.9V_0 < V_0 < 1.1V_0$, V_0 follows from the lattice parameter of Si (5.43 Angstrom)



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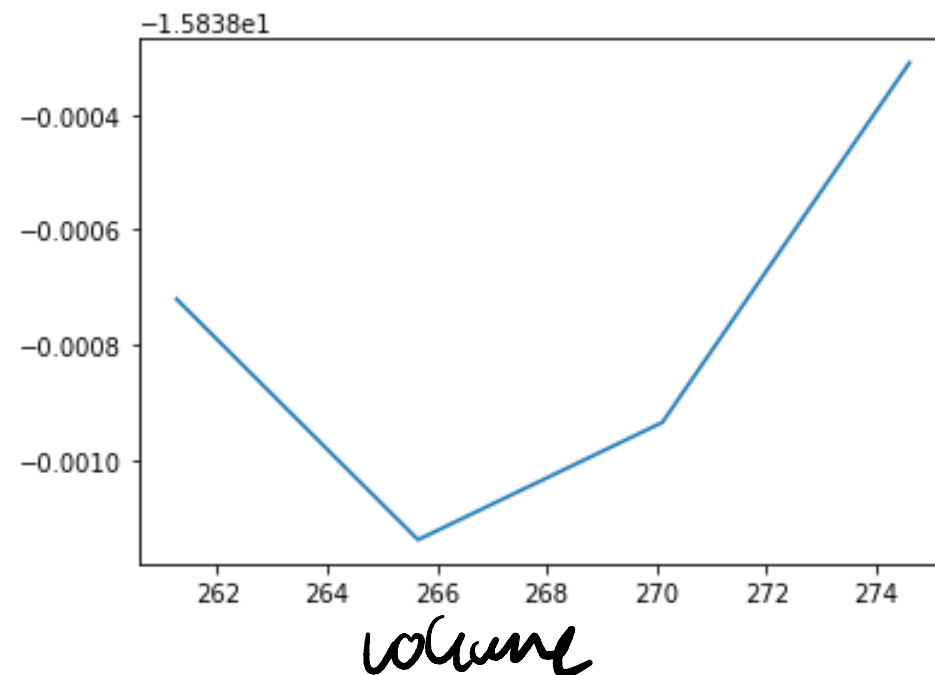
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E_t



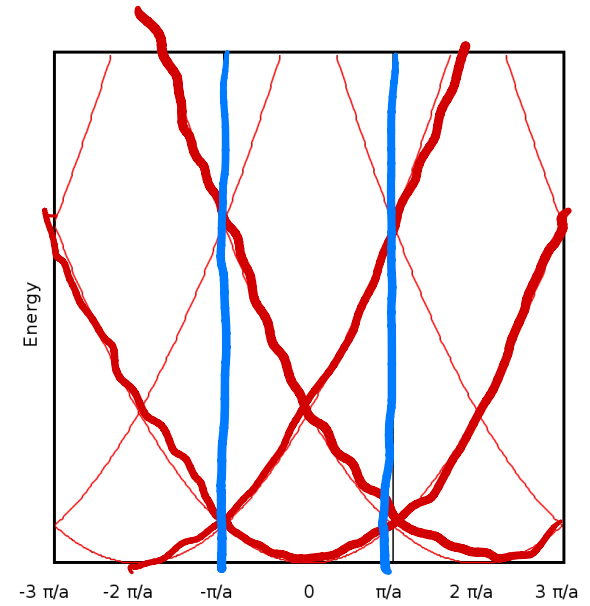
Density Functional Theory: Applications

Example III: Electronic band structure

- Remember from free-electron gas (one-dimensional periodic lattice):

$$E(\mathbf{k}) = \hbar^2 \frac{\mathbf{k}^2}{2m} = \hbar^2 \frac{(\mathbf{k}_{\text{reduced}} + \mathbf{K})^2}{2m}$$

(Note: A red wavy underline is drawn under the \hbar^2 in the second term of the equation.)



(c) https://en.wikipedia.org/wiki/Empty_lattice_approximation

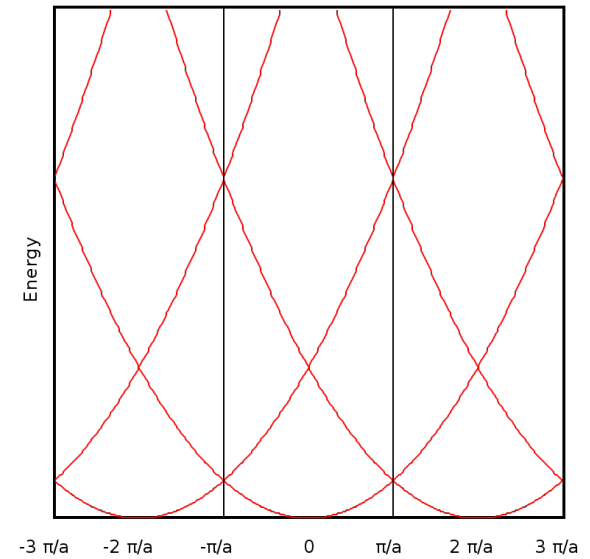
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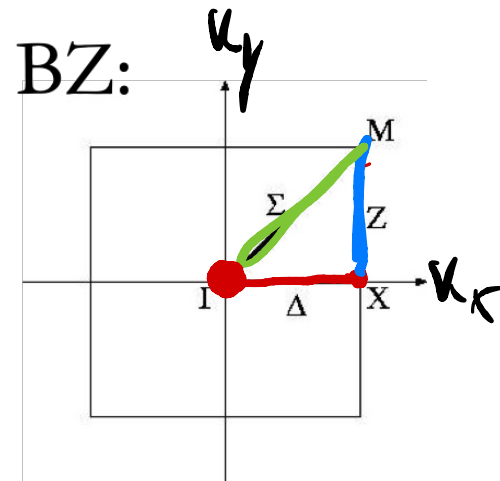
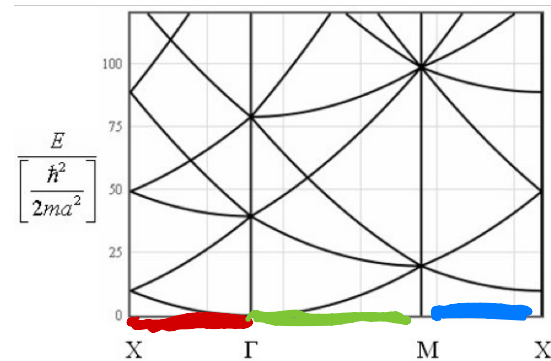
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- in 2D or 3D path in k-space more complicated
- in a real material: electron-ion interaction



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2D (square)



(c) <http://lamp.tu-graz.ac.at/~hadley/ss1/empty/empty.php>

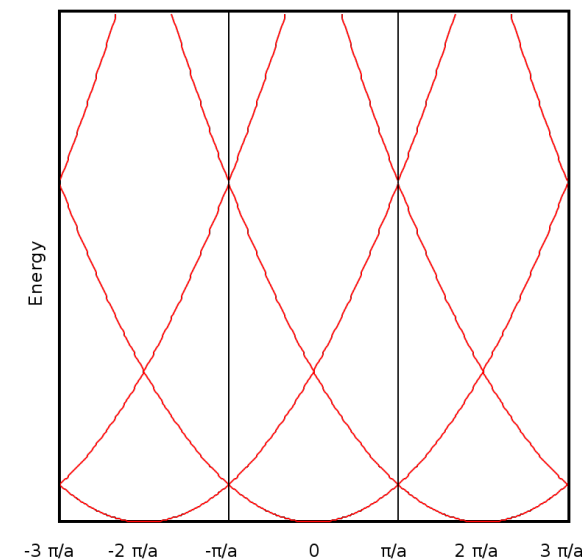
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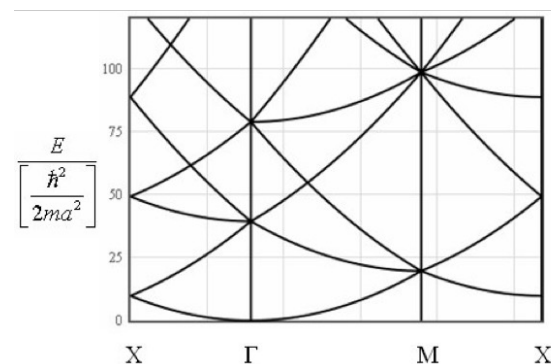
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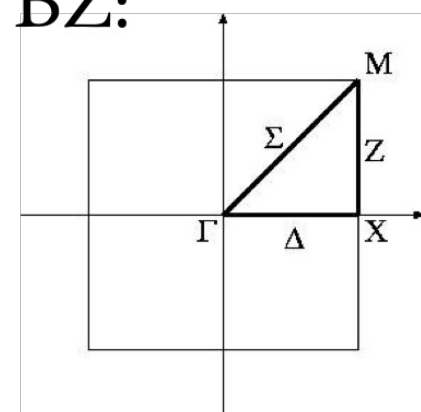
(c) https://en.wikipedia.org/wiki/Empty_lattice_approximation

2D (square)

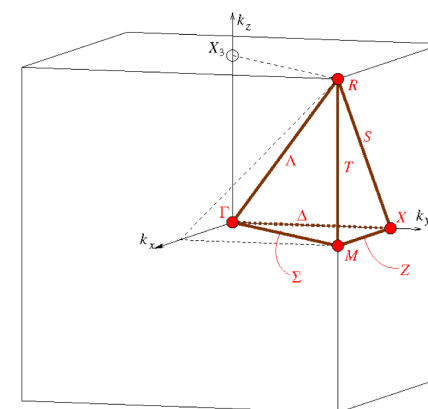


(c) <http://lamp.tu-graz.ac.at/~hadley/ss1/empty/empty.php>

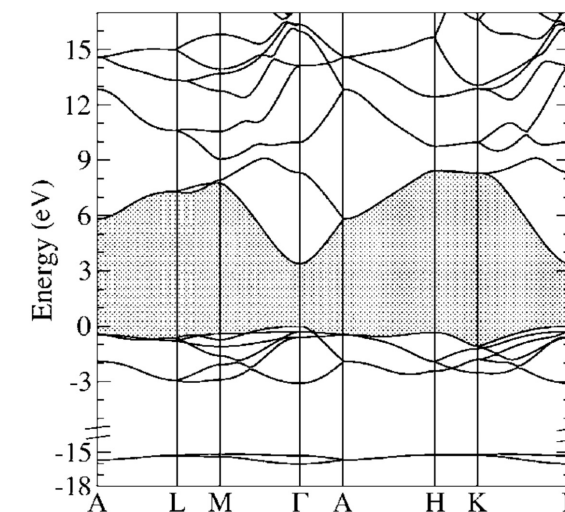
BZ:



3D (schematic)



© bilbao crystallographic server
<http://www.cryst.ehu.es>



(c) Phys. Rev. B 73, 245212 (2006)

Density Functional Theory: Applications

Example III: Electronic band structure

- Restart the DFT tool on nanoHUB
- Use 15 to 20 band structure points
- Pick high-symmetry k-points for your band structure path, e.g. (0.5, 0.5, 0.5); (0.0, 0.0, 0.0); (1.0, 0.0, 0.0)
- Compute the band structure and density of states

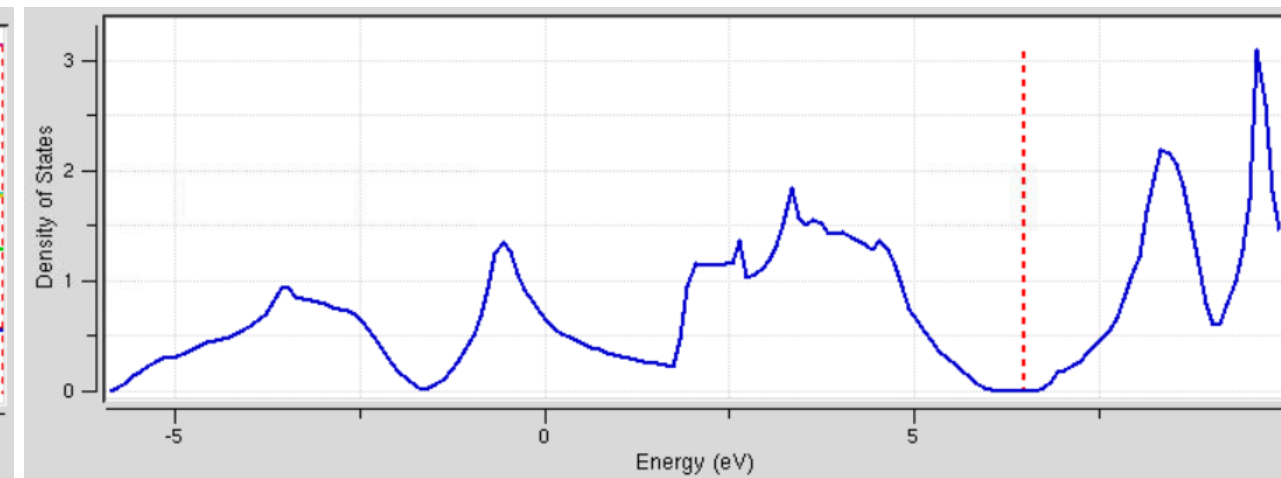
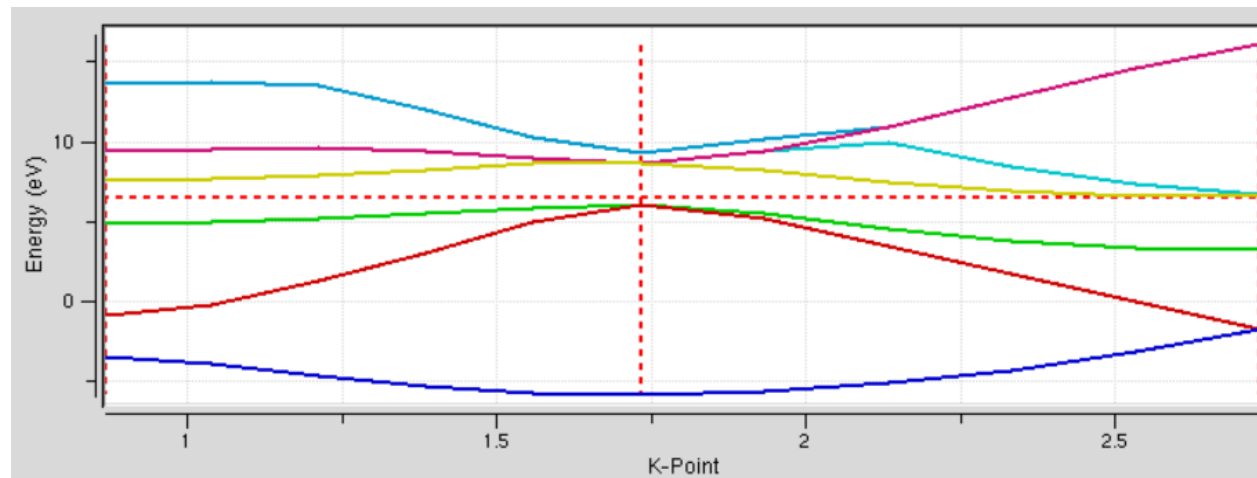


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- Compute the band structure and density of states
- Illustrate the corresponding Brillouin zone path (with the help of the Bilbao Crystallographic Server)
- Label direct and indirect band gaps
- Connect band structure and density of states
- Extract effective masses of electrons and holes



Summary

Literature on Density Functional Theory:

- “Electronic Structure: Basic Theory and Practical Methods”, Richard Martin, Cambridge
- “Density Functional Theory: An Approach to the Quantum Many-Body Problem”, Dreizler and Gross, Springer
- “Density Functional Theory: A Practical Introduction”, David Sholl, Wiley

Literature on Excited Electronic States:

- “Time-Dependent Density-Functional Theory: Concepts and Applications”, Carsten Ullrich, Oxford
- “Many-Body Approach to Electronic Excitations: Concepts and Applications”, Friedhelm Bechstedt, Springer
- “Interacting Electrons: Theory and Computational Approaches”, Reining, Martin, Ceperley, Cambridge



https://nanohub.org/groups/dft_users

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